

⁵⁷Fe Mössbauer Study of Amorphous and Icosahedral $Zr_{65}Al_{7.5}Ni_{10}Cu_{7.3}Fe_{0.2}Ag_{10}$

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(Received July 8, 2000)

The alloys $Zr_{65}Al_{7.5}Ni_{10}Cu_{7.3}Fe_{0.2}Ag_{10}$ in the amorphous and icosahedral state have been studied with ⁵⁷Fe Mössbauer spectroscopy in the temperature range 77–300 K. The average quadrupole splitting decreases with temperature as $T^{3/2}$ and is significantly larger for the icosahedral alloy. The quadrupole splitting in the icosahedral alloy is the largest ever reported for a metallic system. The vibrations of the Fe atoms in both alloys are well described by a simple Debye model.

Introduction

Amorphous (*a*) alloys (glasses) are generally produced from an undercooled liquid state by rapid quenching techniques or quasi-statically at slow cooling. The latter technique has recently led to the development of multicomponent metallic glasses with large glass forming ability and a wide supercooled liquid region before crystallization¹. Very recently, an icosahedral (*i*) phase has been discovered in a Zr–Al–Ni–Cu–Ag metallic glass annealed in a wide supercooled temperature range^{2,4}. Although there are no structural models for the *a*-Zr–Al–Ni–Cu–Ag alloys, it has been speculated² that the structure of these alloys is an assembly of randomly oriented *i* clusters which presumably are also the structural units in *i*-Zr–Al–Ni–Cu–Ag alloys.

This paper reports on a ⁵⁷Fe Mössbauer spectroscopy (MS) study of the *a* and *i* alloys $Zr_{65}Al_{7.5}Ni_{10}Cu_{7.3}Fe_{0.2}Ag_{10}$. It is demonstrated that the hyperfine-interaction parameters are significantly different for these two alloys. It is argued that the structure of the *a* alloy cannot be described in terms of a micro-quasicrystalline model. It is shown that the vibrations of the Fe atoms in both alloys are, surprisingly, well described by a simple Debye model.

Experimental Procedure

An alloy of nominal composition $Zr_{65}Al_{7.5}Ni_{10}Cu_{7.3}Fe_{0.2}Ag_{10}$ was prepared by arc melting in an argon atmosphere of high-purity elemental constituents; the Fe metal used was enriched to 95.9% in the ^{57}Fe isotope. Amorphous ribbons were prepared by melt spinning. The amorphous nature of the ribbons was confirmed with x-ray diffraction and electron microscopy experiments. The differential scanning calorimetry measurements at a heating rate of 0.67 K/s revealed a glass transition temperature at 652 K and two crystallization temperatures at 706 and 769 K associated with two exothermic peaks^{2,3}. The *a* ribbons were annealed at 570 K for 15 min. This resulted in the appearance of nanoscale spherical *i* precipitates, as conformed with the electron microscopy and x-ray diffraction measurements^{2,4}.

^{57}Fe MS measurements were carried out in the temperature range 77–300 K using a standard Mössbauer spectrometer operating in a sine mode⁵. The spectrometer was calibrated with a 6.35- μm Fe foil, and the spectra were folded. The Mössbauer absorbers of the *i* and *a* alloys had the same surface density of 0.043 mg $^{57}Fe/cm^2$. This corresponds to an effective thickness parameter⁵ at 299 K of 0.852 and 0.920 (using the values of the absorber Debye-Waller factors, f_a and f_i , of 0.732 and 0.791 determined below), respectively. As the resulting Mössbauer spectra are due to a multiple of elementary quadrupole doublets, the effective thickness parameter spreads over them and therefore the absorbers can be regarded as being thin⁵.

Results and Discussion

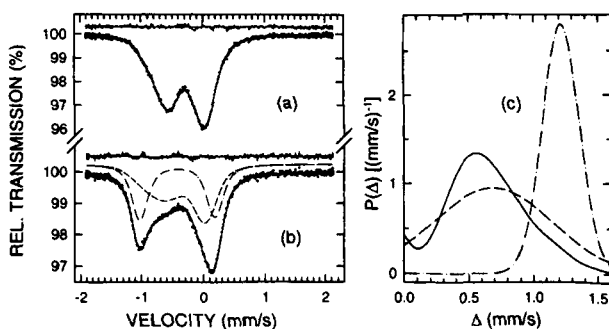


Fig. 1. ^{57}Fe Mössbauer spectra at 299 K of (a) *a* and (b) *i* alloys $Zr_{65}Al_{7.5}Ni_{10}Cu_{7.3}Fe_{0.2}Ag_{10}$ fitted [solid lines in (a) and (b)], respectively, with one $P(\Delta)$ component [solid line in (c)] and two $P(\Delta)$ components [dashed and dot-dashed lines in (c)]. The component fitted spectra are also shown in (b). The residuals are shown above each spectrum.

The Mössbauer spectrum of the *a* alloy consists of a broadened asymmetric doublet [Fig. 1(a)] which results from the distribution of the quadrupole splittings, $P(\Delta)$. It was fitted with the constrained version⁶ of the Hesse-Rübartsch method. An asymmetry of the spectrum was accounted for by assuming a linear relation between the centre shift, $\bar{\delta}$, and Δ of the elementary Lorentzian doublets with width of 0.230 mm/s. A good fit, as judged by the value of $\chi^2=1.15$ and by the residuals [Fig. 1(a)] was obtained for the distribution $P(\Delta)$ shown in Fig. 1(c). It would be desirable to compare this distribution with the one calculated for a structural model of the Zr-Al-Ni-Cu-Ag glasses. Unfortunately, such a model has not been proposed yet for these new *a* alloys.

The Mössbauer spectrum of the *i* alloy [Fig. 1(b)] can only be fitted with two subspectra⁷. The hyperfine parameters and the distribution $P(\Delta)$ of the first subspectrum [Fig. 1(c)] are very close to those corresponding to the *a* alloy. This is consistent with the fact that the *i* sample consists of nanoscale *i* clusters embedded in the *a* matrix. The weight fraction of the *i* phase calculated from the relative area of the subspectrum and the values of f_a and f_i is 41(4)%, which is in agreement with the value obtained from the transmission electron microscopy image of this sample⁷. The second subspectrum results from the *i* phase present in the *i* sample. Its remarkable feature is the very large value of the average quadrupole splitting, $\bar{\Delta}$, which is the largest ever observed for any quasicrystalline alloy⁸ and, to the best of our knowledge, the largest ever found for any metallic alloy. Such a high value of $\bar{\Delta}$ indicates an unusually asymmetric atomic environment around the Cu atoms in the *i* phase. The fact that the distribution $P(\Delta)$ corresponding to the *i* phase is completely different from that corresponding to the *a* phase [Fig. 1(c)] indicates that the structure of *a*-Zr-Al-Ni-Cu-Ag alloys cannot be described in terms of a micro-quasicrystalline model.

The distributions $P(\Delta)$ similar to those in Fig. 1 (c) were determined from the fits of the *a* and *i* samples measured at other temperatures. The temperature dependence of $\bar{\Delta}$ could be fitted [Fig. 2(a)] to the equation $\bar{\Delta}(T) = \bar{\Delta}(0) - BT^{3/2}$, where $\bar{\Delta}(0)$ is the value of $\bar{\Delta}$ at 0 K and B is a constant. Such a temperature dependence has been observed in many noncubic metallic alloys. The values of $\bar{\Delta}(0)$, B determined from the fits for the *a* and *i* phases are respectively 0.6847(20) mm/s, $1.489(88) \times 10^{-5} \text{ K}^{-3/2} \cdot \text{mm/s}$ and 1.2610(29) mm/s, $1.122(69) \times 10^{-5} \text{ K}^{-3/2} \cdot \text{mm/s}$.

The average centre shift at temperature T, $\bar{\delta}(T)$, determined from the fits of the spectra of the *a* and *i* samples is given by $\bar{\delta}(T) = \bar{\delta}_0 + \delta_{\text{SOD}}(T)$, where $\bar{\delta}_0$ is the intrinsic isomer shift and $\delta_{\text{SOD}}(T)$ is the second-order Doppler shift which depends on lattice vibrations of the Fe atoms⁵. In terms of the Debye approximation of the lattice vibrations, $\delta_{\text{SOD}}(T)$ is expressed⁹ by the characteristic Mössbauer temperature Θ_M . By fitting the experimental data

$\bar{\delta}(T)$ [Fig. 2(b)] to the expression $\bar{\delta}(T) = \delta_0 + \delta_{\text{SOD}}(T)$, the quantities δ_0 and Θ_M can be determined. They are $-0.1464(35)$ mm/s, $379(29)$ K and $-0.3034(29)$ mm/s, $439(28)$ K, respectively, for the *a* and *i* samples. The corresponding values of f_a and f_i at 299 K are 0.732(36) and 0.791(24). The different values of δ_0 indicate that electronic properties of the *a* and *i* alloys are very different.

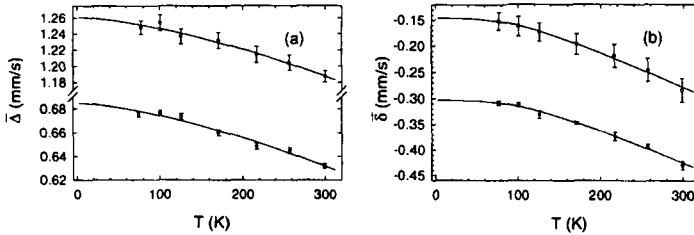


Fig. 2. (a) Temperature dependence of the average quadrupole splitting of the *a* (open circles) and *i* (closed circles) alloys $\text{Zr}_{65}\text{Al}_{7.5}\text{Ni}_{10}\text{Cu}_{7.3}\text{Fe}_{0.2}\text{Ag}_{10}$. The solid lines are the fits, as explained in the text. (b) Temperature dependence of the average centre shift (relative to α -Fe) of the *a* (open circles) and *i* (closed circles) alloys. The solid lines are the fits, as explained in the text.

Acknowledgment This work was supported by the Natural Sciences and Engineering Research Council of Canada.

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