

CRYSTALLINE PROPERTIES OF NICKEL FERRITE-ALUMINATES

J.J. Bara⁺, A.T. Pędziwiatr⁺, Z.M. Stadnik⁺,
A. Szytuła⁺, J. Todorovič⁺⁺, Z. Tomkowicz⁺, W. Zarek⁺⁺⁺

⁺Institute of Physics, Jagiellonian University, Cracow, Poland ; ⁺⁺Boris Kidrič Institute, Vinča, Yugoslavia, ⁺⁺⁺Institute of Physics, Silesian University, Katowice, Poland

The crystalline and magnetic properties of nickel ferrite-aluminates $\text{NiFe}_{2-t}\text{Al}_t\text{O}_4$ ($0 < t < 2$) have been the subject of many investigations performed with different methods. In this paper we report the experimental results of crystalline properties of the $t=0.75, 1.0, 1.25, 1.5, 2.0$ samples obtained with X-ray, neutron diffraction and Mössbauer effect methods. The experimental details are described elsewhere /1/.

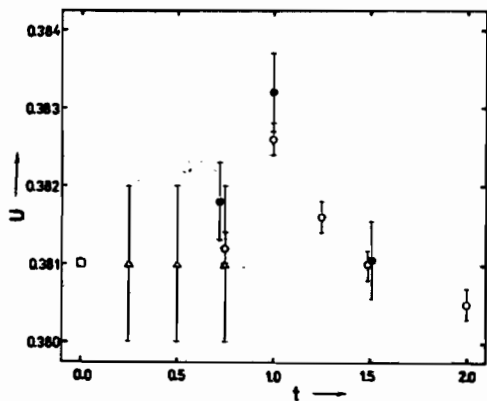


Fig.1. ○ /this work/;

● /2/; □ /3/; △ /4/

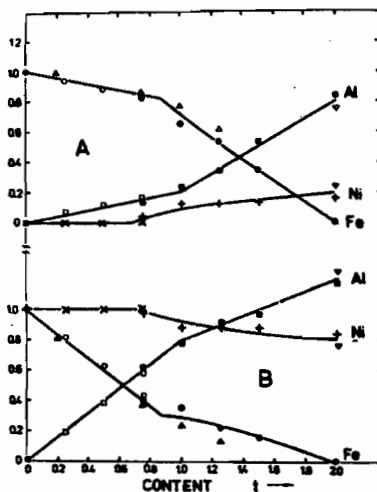


Fig.2. ●, ■, + neutron data /this work/; △, ▲ ME data /this work, 5/; ○, □, × X-ray data /4/; ▽, ▼ X-ray data /6/

The variation of the oxygen parameter u with the sample composition t is shown in Fig.1. Fig.2 shows the experimental and calculated results of the cation distributions for different t values.

The experimental results may be explained with the equilibrium cation distributions model /7/. According to it, for the t values varying from zero to about 0.9 the aluminium cations replace the iron cations in the A and B sites linearly with the probability ratio 1:4. There is no exchange of the nickel cations between the A and B sites. For the t values varying from 0.9 to 2.0 the aluminium cations are embedded into the A and B sites with the probability ratio 3:2. The corresponding ratio for the replacement of the iron cations is slightly higher than 3:2 due to the replacement of the nickel cations from the B sites to the A sites /solid lines in Fig.2/. The cation distributions derived from the experimental data /Fig.2/ are with a good agreement with those predicted by the cation distributions model. There is a difference of the above mentioned ratios at the t value of about 0.9 which should imply the discontinuity of the crystalline properties. Such discontinuity is clearly seen for the oxygen parameter u /Fig.1/.

References:

- /1/ J.J.Bara, A.T.Pędziwiatr, Z.M.Stadnik, A.Szytuła, J.Todorovič, Z.Tomkowicz, W.Zarek: to be published in phys. stat. sol.
- /2/ S.Nizioł, phys. stat. sol. 17a, 555 /1973/
- /3/ J.M.Hastings, L.M.Corliss, Rev. Mod. Phys. 25, 114 /1953/
- /4/ N.M.Chedotaev, A.G.Zalazinskii, V.F.Balakirev, M.T.Kushko, V.M.Bich, G.P.Popov, G.I.Chufarov, Izv. Vyssh. Ucheb. Zaved., Khim. and Khim. Tekhnol. 19, 516 /1976/
- /5/ V.F.Belov, M.N.Shipko, T.A.Khimich, V.V.Korovushkin, L.N.Korablin, Fiz. Tverd. Tela 13, 2018 /1971/
- /6/ F.C.Romeijn, Philips Research Rep. 8, 321 /1953/
- /7/ J.J.Bara: to be published in phys. stat. sol.