Zoned Cr-spinel and ferritchromite alteration in forearc mantle serpentinites of the Rio San Juan Complex, Dominican Republic

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ABSTRACT

Ferritchromite is rarely reported in forearc mantle peridotites. This contribution describes ferritchromite alteration and zoned Cr-spinel in serpentinites from the Rio San Juan Complex in the Dominican Republic. These rocks originated from the forearc mantle and protruded along lithosphere-scale faults in the mid Eocene. The cores of the Cr-spinel grains have Cr# ratios [i.e. atomic Cr/(Cr + Al)] between 0.48 and 0.66; such values are relatively high and are considered to represent primary compositions. Relatively high Zn contents in the grain cores (0.46-0.95 wt.% ZnO) are also thought to be primary; they reflect exceptionally cool conditions in the northern Caribbean forearc mantle. A progressive change in the zoning of Cr-spinel is recorded in the samples. Weakly zoned grains of Cr-spinel have rims with lower Mg# ratios [i.e. atomic Mg/(Mg + Fe²⁺)] and slightly higher Cr# ratios than the cores. More strongly zoned grains of Cr-spinel, in addition to low Mg# and high Cr# in their rims, have a marked increase in Fe^{3+} # [i.e. $Fe^{3+}/(Fe^{3+} + Al + Cr)$] of up to 0.35 in their rims and are partially coated by Mg-rich chlorite. All grains show core-to-rim decreases in their Zn content and increases in Ti, Mn and V. The association with Mg-rich chlorite and the compositional zoning are reminiscent of those reported for ferritchromite. Ferritchromite (with $Fe^{3+\#} > 0.5$) is common in ultramafic rocks in amphibolite-grade terranes; however, the serpentinite samples described herein show little evidence of high-grade metamorphism. The lowtemperature serpentine-group mineral lizardite is dominant and high-temperature antigorite is either very rare or absent; other high-temperature minerals, such as talc, tremolite and cummingtonite, are trace constituents. The observed zoning in the Cr-spinel is thought to represent 'immature' ferritchromite, probably formed in response to a short-lived thermal event. This event appears to have been on too short a timescale to produce either proper ferritchromite or significant quantities of high-temperature minerals. It may be related to the emplacement of the nearby Rio Boba Intrusion, or the upward protrusion of the serpentinites along the lithosphere-scale Septentrional fault zone from the base of the mantle wedge through its hotter interior. We suggest that such alteration is rare in forearc serpentinites because they are not commonly heated during exhumation along the plane of subduction. This work demonstrates that Cr-spinel compositions can be modified by relatively low-grade metamorphism.

Keywords: alteration, serpentinization, zinc, ferritchromite, forearc mantle, Septentrional Fault, exhumation, Hispaniola, northern Caribbean margin.

Introduction

CHROMIUM-SPINEL (chromite) is used as a petrological indicator in a wide variety of mafic and ultramafic rocks (e.g. Irvine, 1967; Dick and Bullen, 1984; Barnes and Roeder, 2001), and is

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especially useful in evaluating the tectonic provenance of serpentinites (e.g. Hattori and Guillot, 2007) because it is commonly the only primary mineral preserved in hydrated peridotites. However, Cr-spinel compositions can be modified during serpentinization and metamorphism. A notable common alteration of Cr-spinel involves an increase in Fe^{3+} and Fe^{2+} contents at the expense of Al³⁺ and Mg²⁺, respectively, without major change in Cr₂O₃ (e.g. Barnes and Roeder, 2001). The resulting high Fe and Cr spinel, informally known in the literature as 'ferritchromite' or 'ferritchromit' (Spangenberg, 1943, and cited in, among others, Evans and Frost, 1975 and González-Jiminez et al., 2009), has been extensively studied and reported from a variety of locations (e.g. Beeson and Jackson, 1969; Ulmer, 1974; Onveagocha, 1974; Bliss and MacLean, 1975; Pinsent and Hirst, 1977; Barnes, 2000;

Standish et al., 2002; Säntti et al., 2006; Merlini et al., 2009; González-Jiminez et al., 2009).

These reported occurrences of ferritchromite are hosted by metamorphosed serpentinites within a variety of tectonic settings. However, the occurrence of ferritchromite in forearc mantle peridotites is very rare and, to our knowledge, has only been reported from hydrated forearc peridotites of the Omi serpentinite mélange in the Hida Mountains of northern Japan (Tsujimori *et al.*, 2004).

This manuscript reports a new occurrence of ferritchromite and zoned Cr-spinel in forearc mantle serpentinites exposed along a major deformation zone in the northern Dominican Republic. These samples record progressive changes in Cr-spinel compositions, which represent the onset of alteration to ferritchromite. Furthermore, these data suggest that there is

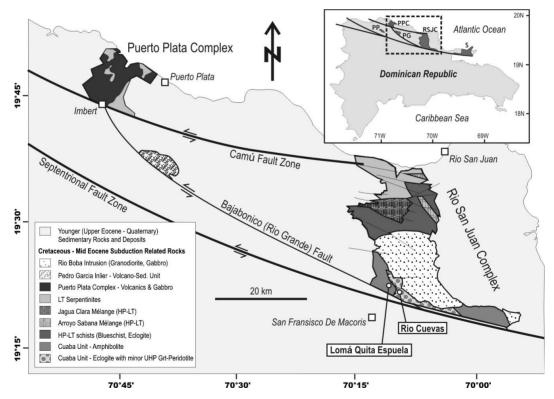


FIG. 1. Map of the investigated area, modified from Draper and Nagle (1991) and Saumur *et al.* (2010). Cuaba serpentinites are found in the Rio Cuevas and Lomá Quita Espuela areas located near the Septentrional Fault Zone (SFZ) northeast of San Fransisco de Macoris. The location of the study area is shown in the inset. Inliers containing subduction related rocks are also shown: S, Samaná; PP, Palma Picada; PG, Pedro Garcia; PPC, Puerto Plata Complex; RSJC, Rio San Juan Complex.

significant mobility of major and trace elements in Cr-spinel even at low degrees of metamorphism. We also speculate on potential reasons for which ferritchromite alteration is uncommon in forearc mantle serpentinites.

Geological setting

Surface exposures in the northern Dominican Republic consist largely of Upper Eocene to Pliocene sedimentary rocks and Quaternary deposits (Lewis et al., 1990). Older rocks crop out in five inliers which expose subductionrelated units of the Rio San Juan complex (RSJC), the Puerto Plata complex, the Pedro Garcia inlier, the Palma Picada inlier and the Samaná Peninsula complex (Fig. 1). These rocks were formed during the subduction of Proto-Caribbean lithosphere (the southern portion of the North American plate) under the northern margin of the Caribbean plate between the late Cretaceous and mid Eocene (e.g. Pindell et al., 2012). Subduction ceased in response to the oblique collision of the Bahamas Platform (North American plate), and this collision resulted in the development of sinistral transform faults along the northern Caribbean margin. The Septentrional and Camú fault zones are two such sinistral transform faults in the northern Dominican Republic (e.g. Mann and Gordon, 1996).

Serpentinites are exposed in the Puerto Plata complex and the RSJC. The Puerto Plata Complex is interpreted to be a deformed ophiolitic complex (Pindell and Draper, 1991) and contains brecciated and sheared serpentinites, tectonized harzburgite, gabbro and volcanic rocks. The RSJC is divided into three components: the northern, central and southern parts. The northern part of the RSJC (also known as the Gaspar Hernandez Serpentinites) is similar to the Puerto Plata complex, and consists of brecciated and massive serpentinites of low metamorphic grade. The central and southern parts of the RSJC expose a fossil subduction channel in which subducted oceanic lithosphere and mantle wedge peridotites physically mingled [according to the interpretations of Gorczyk et al. (2007) and Krebs et al. (2008)]. The central part consists of retrogressed blueschists and eclogites and two serpentinite mélanges (the Jagua Clara and Arroyo Sabana mélanges) which contain metre-scale blocks of blueschist, eclogite and metamorphosed felsic rock aligned with a sub-vertical foliation.

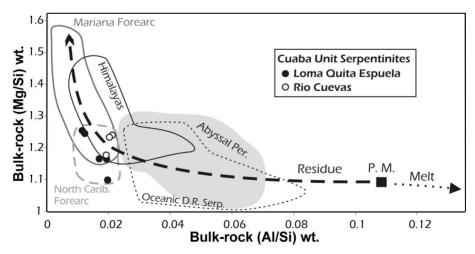


FIG. 2. The Mg/Si vs. Al/Si bulk rock weight ratios of serpentinites in the Cuaba Unit, modified from Saumur *et al.* (2010). The compositional variations expected during progressive partial melting are shown by the arrows, and primitive mantle values (P.M.) are taken from McDonough and Sun (1995). Bulk-rock compositions of Cuaba Unit serpentinites reported by Saumur *et al.* (2010) are consistent with a forearc mantle origin. Data sources are as follows: Mariana forearc serpentinites (Ishii *et al.*, 1992; Parkinson and Pearce, 1998); serpentinites from Talnakh, Himalayas (Guillot *et al.*, 2001); forearc serpentinites from the northern Caribbean margin (Bowin *et al.*, 1966; Hattori and Guillot, 2007; Saumur *et al.*, 2010); and abyssal peridotite (Abyssal Per., Niu, 2004; Oceanic DR Serp, Saumur *et al.*, 2010).

Retrogressed eclogitic rocks originating from oceanic protoliths (Hattori *et al.*, 2010*a*) are the dominant rocks in the southernmost unit of the RSJC and are described herein as the Cuaba Unit [these are described as the Cuaba Amphibolite in some earlier publications (e.g. Draper and Nagle, 1991)]. Small amounts of ultra high pressure (UHP) garnet peridotite have been reported along the Rio Cuevas stream (Fig. 1); the origin of these UHP rocks is currently a matter of debate (e.g. Abbott and Draper, 2010; Hattori *et al.*, 2010*a,b*; Gazel *et al.*, 2011; De Hoog, 2012; Gazel *et al.*, 2012).

Serpentinites represent only minor components of the eclogitic Cuaba Unit. They form lenticular bodies within the unit (Hattori *et al.*, 2010*a*; Saumur *et al.*, 2010) which crop out ~10 km northeast of San Fransisco de Macoris (Fig. 1). In the Lomá Quita Espuela area, the serpentinites form 100 m scale exposures along gravel roads. They are generally massive and locally exhibit a steeply dipping foliation of moderate intensity. The serpentinites are also abundant as large angular boulders, up to ~1 m in size, along the Rio Cuevas. Although they were not located *in situ*, the large size, shape and the abundance of these boulders in a narrow (5–10 m) and shallow (<1 m) stream suggest a proximal source.

An analysis of whole-rock geochemistry and primary Cr-spinel compositions by Saumur et al. (2010) showed that two types of serpentinite occur in the northern Dominican Republic. The first type is present in large volumes within the subduction complex and occurs in the Puerto Plata Complex, Gaspar Hernandez Serpentinites, Jagua Clara Mélange and Arroyo Sabana Mélange. These serpentinites have high Al and low Mg bulk-rock compositions and contain relatively chromium-poor Cr-spinel; these characteristics are consistent with an abyssal peridotite origin. The second type of serpentinite is interpreted as originating from forearc mantle peridotites. It occurs along the major Septentrional and Camú strike-slip fault zones, and has high Mg and low Al bulk-rock compositions (Fig. 2) and contains chromiumrich Cr-spinel (Fig. 3). On the basis of the distribution of forearc serpentinites and the observed deformation fabrics within the serpentinites and host rocks, the forearc serpentinites are considered to have protruded from the base of a mantle wedge to shallow crustal levels along fault zones during post-Eocene transpression (Saumur et al., 2010). These relatively buoyant serpenti-

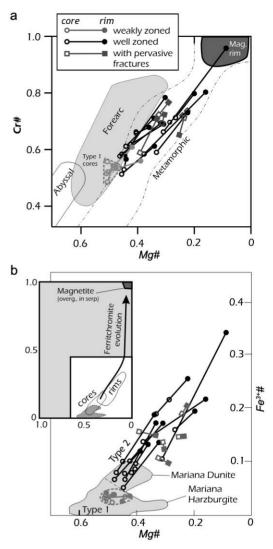


FIG. 3. Composition of cores and rims of zoned Cr-spinels; (a) Cr# vs. Mg# and (b) $Fe^{3+}\#$ vs. Mg#. Open circles and squares represent the cores of grains and filled symbols represent the rims. One core-rim pair represents one grain. The legend remains the same for subsequent figures. The forearc field is defined by spinel in peridotites from the Mariana Forearc (Ishii *et al.*, 1992) and the abyssal peridotite field is after Dick and Bullen (1984). The field of metamorphic Cr-spinels was redrawn from Säntti *et al.* (2006) after Evans and Frost (1975).

nites probably contributed to the exhumation of HP and UHP rocks in the subduction channel (Gorczyk *et al.*, 2007; Hattori *et al.*, 2010*a*).

The RSJC includes the Rio Baba Intrusion (Fig. 1), which has not been the focus of extensive study. It consists of granodiorite and gabbro and is thought to have intruded the Cuaba Unit (Draper and Nagle, 1991). The timing of its emplacement

relative to regional faulting is not known, but intrusion contacts appear to be at least partly faultcontrolled, suggesting that emplacement was relatively late. No obvious contact aureole has been observed, but its emplacement may, never-

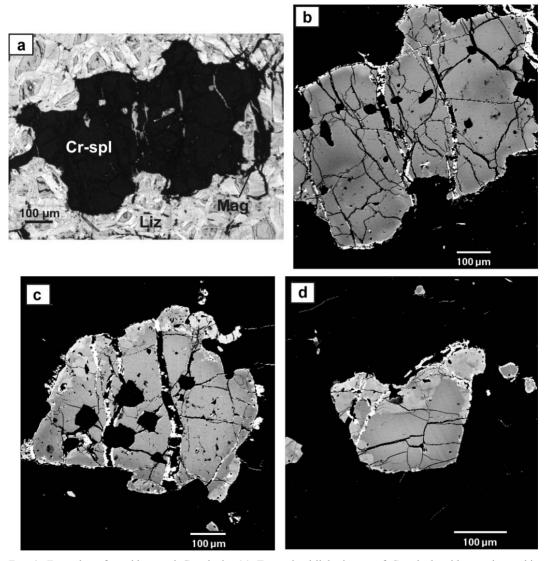


FIG. 4. Examples of weakly-zoned Cr-spinels. (a) Transmitted-light image of Cr-spinel, with pseudomorphic lizardite (Liz) forming the bulk of the serpentinite groundmass and minor magnetite (Mag) located at grain edges. (b) Backscattered-electron image of the grain shown in Fig. 3a. Darker zones within the Cr-spinel grain have slightly higher Cr#. Note the bright magnetite in the cracks. (c) Backscattered-electron image of Cr-spinel, generally not zoned except for patchy reflective zones near rims (upper part and lower right side of grain) which are due to lower Mg#. Highly reflective magnetite overgrowths occur in fractures within grains. The black material is lizardite. The embayed nature of the grain suggests a primary morphology. (d) Backscattered-electron image of weakly zoned Cr-spinel with slightly higher Fe in the rim. Magnetite fills cracks and surrounds the upper right corner of the spinel.

theless, have caused slight metamorphism of neighbouring Cuaba Unit.

This contribution focusses exclusively on samples of forearc mantle serpentinites from the Cuaba Unit (collected along the Rio Cuevas and at Lomá Quita Espuela); it is these serpentinites which exhibit various degrees of compositional zoning of Cr-spinel.

Analytical methods

Serpentinite samples were collected along the Rio Cuevas (samples RD45, RD48 and RD60) and at Lomá Quita Espuela (samples RD34a, RD34c, RD68, RD6-36a and RD6-36c). The Cr-spinels were examined by transmitted-light and incidentlight microscopy. Backscattered-electron (BSE) images were collected using a Lamont four element solid-state detector and BSE Quad Summing Amplifier. Mineral compositions (27 grains) were determined on an automated CAMEBAX MBX electron microprobe using four wavelength-dispersive spectrometers. Counting times were 15 s per element, except for Fe (20 s) and Ni (40 s). A 15 kV accelerating potential and 20 nA beam current were used. The calibration standards were pure wollastonite (Si. Ca), synthetic spinel (Al), synthetic Cr₂O₃ (Cr), forsterite (Mg), synthetic MnTiO₃ (Mn, Ti), pure vanadium metal (V), albite (Na), fayalite (Fe in silicates), synthetic Fe₂O₃ (Fe in oxides), tugtupite (Cl) and synthetic lithium fluoride (F).

The Fe_2O_3 content of the Cr-spinel was calculated assuming stoichiometric compositions (e.g. Bliss and MacLean, 1975; Dick and Bullen, 1984; Barnes and Roeder, 2001). One drawback of this method is that errors associated with other

elements propagate into larger errors in estimated Fe^{3+} (Wood and Virgo, 1989). This is important to recognize when dealing with relatively small variations in ferric iron content (Barnes and Roeder, 2001). However, we consider this effect to be negligible in our samples (Table 1) because Fe_2O_3 values are generally high (>2 wt.%), and intragrain variations are significant (>10 wt.%).

Serpentine-group minerals were identified using powder X-ray diffraction and micro-Raman spectrometry. The analytical methods pertaining to these are described by Saumur *et al.* (2010).

Results

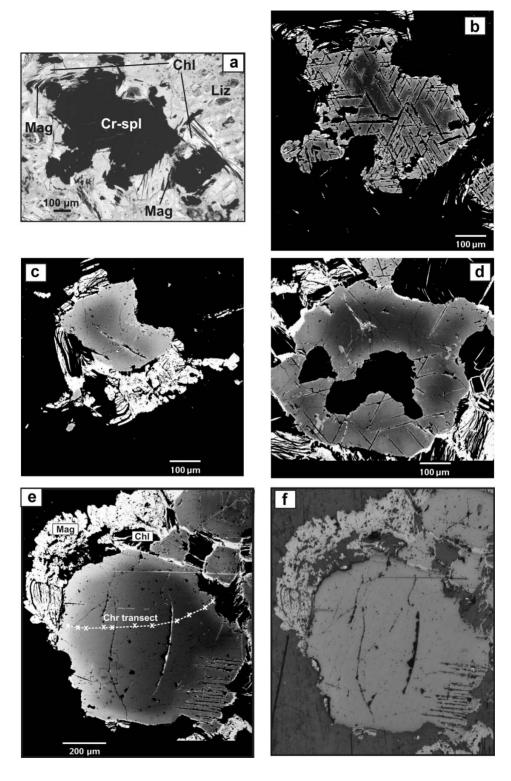
Petrography of the Cuaba serpentinites

All of the samples are almost fully hydrated (11-14% loss on ignition) and are dominated by pseudomorphic mesh-textured serpentine after olivine (Saumur et al., 2010). Bastite after orthopyroxene is rarely observed. Samples RD34a and RD68 are the only ones in which minor amounts (<1 vol.%) of primary olivine (Fo₉₀) and orthopyroxene (En₈₉₋₉₁) remain; in most samples Cr-spinel is the only preserved primary mineral. Fine-grained dusty magnetite forms less than 1 vol.% of samples and has nearendmember compositions. Samples RD34 and RD6-36 contain minor talc and tremolite (<2 vol.%) and traces of cummingtonite. No secondary metamorphic olivine and pyroxene were found.

Powder X-ray diffraction (XRD), micro-Raman spectrometry and microscopy suggest that the dominant serpentine-group mineral is lizardite, with minor chrysotile (Saumur *et al.*, 2010). The

FIG. 5 (*facing page*). Examples of well zoned Cr-spinels. (*a*) Transmitted-light photomicrograph of the Cr-spinel grain in plane-polarized light. Note the chlorite (Chl, clear, high relief) around the Cr-spinel (Cr-spl) which overprints lizardite (Liz) and is intergrown with magnetite (Mag) (Sample RD-34c). Magnetite has an endmember composition, similar to dusty magnetite disseminated in lizardite. (*b*) Backscattered-electron image of the Cr-spinel shown in Fig. 4*a*; Note the polygonal pervasive fractures (relicts of trellis exsolution or partings) and brighter rims due to higher Fe^{3+} # than cores. (*c*) Backscattered-electron image of Cr-spinel showing gradual compositional zoning contrast with magnetite overgrowths. Magnetite is intergrown with chlorite (black), and the matrix is lizardite (also black) (Sample RD48). (*d*) Backscattered-electron image of Cr-spinel with diffuse zoning, surrounded by overgrowths of symplectitic magnetite (bright) and chlorite (dark). Dark areas in the cores of the grain are aggregates of chlorite and lizardite. Note the minor microscopic fractures in the lower portion of the grain (Sample RD48). (*e*) Backscattered-electron image of a Cr-spinel grain showing gradual zoning, partly surrounded by intergrowths of magnetite and chlorite. Composition variation along the transect ('Chr-transect') is shown in Fig. 6. Note the minor microscopic fractures at the bottom right corner of the grain. (*f*) Incident light photomicrograph of the grain shown in Fig. 4*e* under open polars. Note the similar intragrain reflectivity of Cr-spinel in contrast to the high reflectivity of magnetite surrounding the grain, which suggest that they are two distinct phases.

ZONED CR-SPINEL AND FERRITCHROMITE ALTERATION



$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$			Weakly	Weakly zoned					Well zoned				With p	With pervasive
once Cr-spinel tim Cr-spinel core Cr-spinel tim Cr-spinel tim-spinel tim-spinel <tht< th=""><th>Sample</th><th>— 68</th><th>B-4 —</th><th>-68_{i}</th><th>A-1 —</th><th></th><th>– RD48-T -</th><th></th><th>- RD3</th><th>4C-1 —</th><th>– RD3</th><th>4A-K –</th><th>- RD</th><th>1actures 60-2 —</th></tht<>	Sample	— 68	B-4 —	-68_{i}	A-1 —		– RD48-T -		- RD3	4C-1 —	– RD3	4A-K –	- RD	1actures 60-2 —
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		core Cr-spinel	rim Cr-spinel	core Cr-spinel	rim Cr-spinel	core Cr-spinel	rim ferritchr.	magnetite overgrowth	core Cr-spinel	rim ferritchr.	core Cr-spinel	rim ferritchr.	core ferritchr.	rim ferrtichr.
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	SiO_2	I	I	I	I	0.04	0.12	0.81	I	I	0.52	0.13	I	0.06
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Al_2O_3	25.26	24.61	21.28	19.78	20.71	13.22	0.03	23.24	20.06	15.22	1.21	12.3	10.49
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	TiO_2	0.02	0	0.03	0.07	0.06	0.23	I	0.08	0.14	0.15	0.95	0.28	0.17
40.73 12.74 41.31 45.90 43.94 44.36 0.82 40.31 38.28 93.29 44.83 45.94 43.14 61.96 63.24 63.15 85.69 53.75 53.69 53.75 53.69 53.75 53.69 53.75 53.69 53.75 53.69 53.75 53.69 53.75 100.79 10 0.33 100.65 100.27 99.75 90.75 0.71 0.45 0.47 0.45 0.47 0.45 0.47 0.46 107.79 10 0.33 100.65	V_2O_3	0.25	0.3	0.27	0.3	0.16	0.23	Ι	0.28	0.32	0.25	0.32	0.55	0.44
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Cr_2O_3	40.73	42.74	44.31	45.99	43.99	44.36	0.82	40.31	38.28	44.68	41.86	48.21	50.86
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$^{\dagger}\mathrm{Fe_{2}O_{3}}$	3.62	2.15	4.05	3.02	4.48	10.65	65.58	5.69	9.7	8.33	23.71	8.39	7.56
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	FeO	18.85	19.82	19.47	21.2	20.11	23.44	31.54	19.5	21.87	24.54	29.26	23.69	24.51
$ \begin{array}{ccccccccccccccccccccccccccccccc$	MgO	10.78	10.08	9.83	8.62	9.25	6.32	0.12	9.93	7.86	6.19	1.58	6.33	5.67
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	MnO	0.31	0.31	0.52	0.25	0.27	0.42	0.05	0.28	0.37	0.46	0.79	0.58	0.45
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	NiO	n.a.	n.a.	n.a.	n.a.	0.07	0.12	I	0.1	0.14	0.14	0.18	n.a.	n.a.
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	ZnO	0.53	0.64	0.51	0.55	0.82	0.71	Ι	0.77	0.68	0.95	0.47	0.46	0.47
100.35 100.65 100.27 99.78 99.96 99.82 98.95 100.18 99.42 101.43 100.46 100.79 10 - - - - - - - 0.14 0.04 - - - - - - - - 0.01 0.03 0.25 - - 0.14 0.04 - 3.88 0.00 0.00 0.01 0.03 0.05 - 0.01 0.03 0.02 0.01 0.03 0.02 0.01 0.03 0.02 0.01 0.03 0.02 0.01 0.03 0.02 0.01 0.03 0.0	CaO	Ι	Ι	I	Ι	Ι	I	Ι	I	I	I	I	I	I
itions on the basis of 32 oxygen atoms $\begin{array}{cccccccccccccccccccccccccccccccccccc$	Total	100.35	100.65	100.27	99.78	96.66	99.82	98.95	100.18	99.42	101.43	100.46	100.79	100.68
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Composi	itions on the	basis of 32		ns									
7.34 7.18 6.32 5.99 6.21 4.19 0.01 6.86 6.13 4.70 0.42 3.88 0.00 0.00 0.01 0.01 0.01 0.05 - 0.01 0.03 0.03 0.21 0.06 0.05 0.06 0.05 0.06 0.03 0.05 - 0.06 0.07 0.05 0.08 0.12 7.93 8.36 8.84 9.34 8.85 9.44 0.20 7.98 7.85 9.26 9.75 1020 0.66 0.06 0.11 0.58 0.86 2.16 15.29 1.07 1.89 1.65 5.26 1.69 3.96 3.72 3.70 3.30 3.51 2.53 0.06 3.71 3.04 2.42 0.70 2.53 0.06 0.06 0.01 0.01 0.01 0.06 0.08 0.10 0.01 0.06 0.13 0.03 0.01 0.01 0.06 0.08 0.10 0.01 0.03 0.10 0.12 0.11 0.05 0.06 0.14 - 0.14 0.13 0.01 0.00 0.01 0.10 0.12 0.10 0.10 0.15 0.14 - 0.14 0.13 0.01 0.03 0.03 0.04 n.a. 0.10 0.12 0.010 0.15 0.14 0.14 0.13 0.18 0.10 0.09 0.10 0.12 0.010 0.010 0.07 0.05 0.03 0.04 n.a. 0.10 0.12 0.04 0.05 0.04 0.05 0.14 0.13 0.18 0.10 0.09 0.52 0.54 0.58 0.61 0.59 0.69 0.96 0.54 0.56 0.66 0.96 0.72 0.04 0.03 0.04 0.05 0.14 0.13 0.11 0.34 0.11 0.39 0.50 0.04 0.03 0.01 0.010 0.012 0.01 0.09 0.51 0.54 0.58 0.61 0.59 0.69 0.96 0.54 0.56 0.66 0.96 0.72 0.50 0.04 0.03 0.05 0.04 0.05 0.14 0.13 0.11 0.34 0.11 0.39 0.50 0.04 0.03 0.50 0.04 0.05 0.14 0.99 0.07 0.12 0.11 0.34 0.11 0.50 0.66 0.96 0.96 0.54 0.56 0.66 0.96 0.72 0.50 0.66 0.96 0.96 0.54 0.58 0.61 0.59 0.69 0.90 0.07 0.12 0.11 0.34 0.11 0.54 0.56 0.66 0.96 0.96 0.54 0.58 0.50 0.96 0.72 0.91 0.48 0.93 0.31 0.09 0.32 0.50 0.54 0.58 0.54 0.58 0.54 0.56 0.56 0.56 0.56 0.56 0.56 0.56 0.56	Si	I	I	1	I	0.01	0.03	0.25	I	I	0.14	0.04	I	0.06
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Al	7.34	7.18	6.32	5.99	6.21	4.19	0.01	6.86	6.13	4.70	0.42	3.88	3.35
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Ti	0.00	0.00	0.01	0.01	0.01	0.05	Ι	0.01	0.03	0.03	0.21	0.06	0.04
7.938.368.849.348.859.440.207.987.859.269.7510.200.670.400.770.580.862.1615.291.071.891.655.261.693.884.104.114.554.285.288.174.084.755.387.215.303.963.723.703.303.512.530.060.100.000.100.702.533.963.723.703.303.512.530.060.110.050.072.530.060.060.110.050.060.100.000.010.020.790.100.120.110.050.060.14-0.140.130.100.200.100.120.100.100.030.04n.a.0.100.120.100.140.520.540.580.610.590.690.960.540.560.660.960.750.640.030.040.590.690.990.070.120.100.090.050.640.030.040.590.690.960.540.560.660.960.720.700.330.030.050.140.990.070.120.110.09 </td <td>Λ</td> <td>0.05</td> <td>0.06</td> <td>0.05</td> <td>0.06</td> <td>0.03</td> <td>0.05</td> <td>I</td> <td>0.06</td> <td>0.07</td> <td>0.05</td> <td>0.08</td> <td>0.12</td> <td>0.10</td>	Λ	0.05	0.06	0.05	0.06	0.03	0.05	I	0.06	0.07	0.05	0.08	0.12	0.10
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Cr	7.93	8.36	8.84	9.34	8.85	9.44	0.20	7.98	7.85	9.26	9.75	10.20	10.91
3.88 4.10 4.11 4.55 4.28 5.28 8.17 4.08 4.75 5.38 7.21 5.30 3.96 3.72 3.70 3.30 3.51 2.53 0.06 3.71 3.04 2.42 0.70 2.53 0.06 0.06 0.11 0.05 0.06 0.10 0.01 0.06 0.08 0.10 0.20 0.13 n.a. n.a. n.a. n.a. 0.01 0.03 - 0.02 0.03 0.04 n.a. 0.10 0.12 0.10 0.10 0.15 0.14 - 0.14 0.13 0.18 0.10 0.09 	Fe^{3+}	0.67	0.40	0.77	0.58	0.86	2.16	15.29	1.07	1.89	1.65	5.26	1.69	1.54
3.96 3.72 3.70 3.30 3.51 2.53 0.06 3.71 3.04 2.42 0.70 2.53 0.06 0.06 0.11 0.05 0.06 0.10 0.01 0.06 0.08 0.10 0.20 0.13 n.a. n.a. n.a. n.a. 0.01 0.03 - 0.02 0.03 0.04 n.a. 0.10 0.12 0.10 0.10 0.15 0.14 - 0.14 0.13 0.18 0.10 0.09 	Fe^{2+}	3.88	4.10	4.11	4.55	4.28	5.28	8.17	4.08	4.75	5.38	7.21	5.30	5.56
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Mg	3.96	3.72	3.70	3.30	3.51	2.53	0.06	3.71	3.04	2.42	0.70	2.53	2.29
n.a. n.a. n.a. n.a. n.a. 0.01 0.03 - 0.02 0.03 0.04 n.a. 0.0 0.10 0.12 0.10 0.10 0.15 0.14 - 0.14 0.13 0.18 0.10 0.09 	Mn	0.06	0.06	0.11	0.05	0.06	0.10	0.01	0.06	0.08	0.10	0.20	0.13	0.10
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Ż	n.a.	n.a.	n.a.	n.a.	0.01	0.03	I	0.02	0.03	0.03	0.04	n.a.	n.a.
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Zn	0.10	0.12	0.10	0.10	0.15	0.14		0.14	0.13	0.18	0.10	0.09	0.09
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Са	Ι	Ι	I	I	I	I	I	I	I	I	I	Ι	Ι
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Cr#*	0.52	0.54	0.58	0.61	0.59	0.69	0.96	0.54	0.56	0.66	0.96	0.72	0.76
2 0.01 0.48 0.39 0.31 0.09 0.32 + Al); $Mg\# = Mg/(Mg + Fe^{2+})$.	$Fe^{3+}\#*$	0.04	0.03	0.05	0.04	0.05	0.14	0.99	0.07	0.12	0.11	0.34	0.11	0.10
* The ratios are defined as: $Cr\# = Cr/(Cr + AI)$; $Fe^{3^+}\# = Fe^{3^+}/(Fe^{3^+} + Cr + AI)$; $Mg\# = Mg/(Mg + Fe^{2^+})$.	$Mg^{\#*}$	0.50	0.48	0.47	0.42	0.45	0.32	0.01	0.48	0.39	0.31	0.09	0.32	0.29
	* The r	atios are del	fined as: <i>Cr</i> #	i = Cr/(Cr +	. Al); Fe ³⁺ #	$= Fe^{3+}/(Fe^{3+})$	$^{+}$ + Cr + Al	(); <i>Mg</i> # = Mg	$/(Mg + Fe^{2+})$					

TABLE 1. Representative compositions of Cr-spinel and magnetite from the Rio San Juan Complex.

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pseudomorphic mesh texture of lizardite suggests that olivine was converted to lizardite, which is typical of low-temperature serpentinization (O'Hanley, 1996). The high-temperature serpentine-group mineral antigorite (e.g. Evans, 2004), was not detected by XRD or micro-Raman spectrometry. Thus, the mineralogy and textural relationships suggest that the Cuaba serpentinites have undergone a relatively low degree of heating. Lizardite grains analysed in this study have variable total Fe (2.65-11.22 wt.%) and Al (0.03-1.18 wt.% Al₂O₃) contents.

Chromium-spinel

Chromium-spinel constitutes less than 1 vol.% of the serpentinites. Grain sizes vary within samples, some are as small as 50 μ m, but most are between 0.3 and 1.2 mm. Grains are subhedral-to-anhedral, and rarely are vermicular or embayed. They appear dark red to dark brown (Fig. 4*a*) or opaque in transmitted light (Fig. 5*a*). Their opacity appears to be broadly correlated with their Fe³⁺ content.

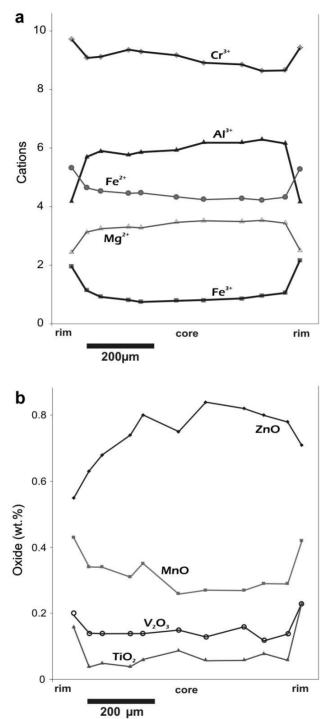
The compositions of the cores of Cr-spinel grains are similar within individual samples. They are characterized by moderate Cr# [i.e. atomic Cr/(Cr + Al) ratios of 0.48-0.66 and moderate Mg# [i.e. atomic Mg/(Mg + Fe²⁺)] ratios of 0.40-0.51 (Table 1; Fig. 3a). Most grains, with some exceptions, have low Fe^{3+} # [i.e. atomic $Fe^{3+}/(Cr + Al + Fe^{3+})$] ratios of 0.03-0.10, which are within the range expected for mantle-wedge peridotites (e.g. Ishii et al., 1992; Ohara and Ishii, 1998; Fig. 3b). These features, along with low Ti (<0.21 wt.% TiO₂) contents, suggest that the grain cores are not significantly metamorphosed, and probably represent primary compositions. The Zn content in the Cr-spinel cores varies from 0.46 to 0.95 wt.% ZnO, which is unusually high for Crspinels in mantle-wedge peridotites (see compilation by Arai and Ishimaru, 2011).

On the basis of the observed textural and compositional changes in the Cr-spinel, a sequence of events can be established. Chromium-spinel grains from sample RD68 are weakly zoned (and are termed 'weakly zoned' in Fig. 3 and subsequent figures). The edges of these grains are slightly brighter in backscattered-electron (BSE) images (Fig. 4b-d) as a result of an increase in their Fe²⁺ content and a decrease in Mg²⁺. However, zoning is patchy and a concentric pattern is not well developed (Fig. 4c, top left side of grain; Fig 4d, top part of grain). The *Cr#* ratios

are slightly higher at the grain rims, as shown by the slightly darker zones in close proximity to the bright rims in the BSE images (Fig. 4c-d), and in diffuse darker areas within the grains (Fig. 4b). The Fe³⁺ content at the rims remains low and is similar to that at the cores (Table 1).

The Cr-spinel in all of the other samples described in this study (RD34a, RD34c, RD6-36, RD45, RD48 and RD60) exhibits more prominent compositional zoning (termed 'well zoned' in Fig. 3 and subsequent figures) than sample RD68. In transmitted light, such grains are either nearly opaque or have dark brown cores which gradually increase in opacity toward their rims (Fig. 5a). Gradational concentric compositional zoning is apparent in BSE images (Fig. 5c-e). In addition to a decrease in Mg#and a small increase in Cr content, a pronounced increase in Fe^{3+} # occurs at the rims (Figs 3 and 6a). Rim compositions plot close to the low Fe end of the ferritchromite composition field (Figs 3b and 7) which is defined by published data on ferritchromite in ultramafic rocks metamorphosed under amphibolite facies conditions reported from other localities (e.g. Evans and Frost, 1975; Pinsent and Hirst, 1977; Liipo et al., 1995b; Suita and Strieder, 1996; Barnes and Roeder, 2001). Overall, the rims of well zoned grains have lower Al. Mg. and Zn contents and are enriched in Fe, Cr, Ti, Mn, Ni and V compared to the cores (Table 1; Figs 6-8).

Several grains of Cr-spinel in samples RD60 and RD34a (Fig. 5a,b) are cut by numerous microscopic fractures which intersect at constant angles within a given grain. These fractures are commonly filled with clinochlore or lizardite. Although such fractures might represent the octahedral spinel partings described by Deer et al. (1992), it is also possible that they represent relics of lattice exsolution along {111} crystallographic planes. Such exsolution has been reported to occur in Cr-spinel during oxidation (Haggerty, 1991). These grains have similar compositions to the well zoned grains described above ('with pervasive fractures' in Fig. 3 and in subsequent figures), but have partly or completely lost the zoning pattern observed in the well zoned grains (Figs 3 and 5*b*). The Fe^{3+} and Fe^{2+} contents adjacent to fractures (Fig. 5c) and possible exsolution relics (Fig. 5b,d,e) are commonly elevated, and show up as bright areas in BSE images, extending up to 50 µm away from the fractures. Furthermore, the compositions of grains with pervasive microscopic fractures plot just



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FIG. 6. Compositional variation along the transect of the zoned Cr-spinel grain shown in Fig. $4e_f$ (from sample RD 48). The Fe³⁺ contents are calculated assuming stoichiometric composition of Cr-spinel. (*a*) Major elements and (*b*) minor elements.

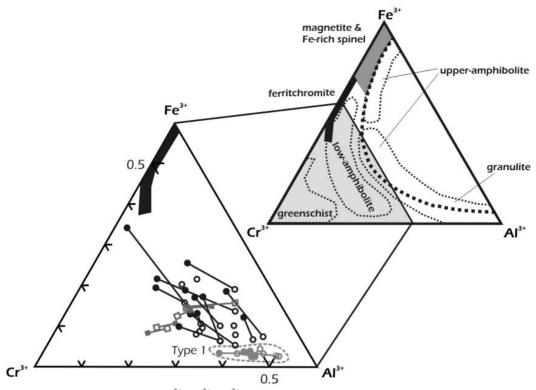


FIG. 7. Ternary diagram of atomic Cr³⁺-Fe³⁺-Al³⁺ compositions. As in Fig. 3, the magnetite in the Cuaba samples is low in Cr and plots in the field of 'mag rims'. The field for ferritchromite and its compositional trend was obtained from data and diagrams given by Barnes and Roeder (2001), Pinsent and Hirst (1977) and Liipo *et al.* (1995b). Fields for greenschist, lower-amphibolite, upper-amphibolite and granulite-grade spinels were obtained from Evans and Frost (1975) and Suita and Strieder (1996), as cited by González-Jiménez *et al.* (2009).

outside the fields of forearc peridotite (Fig. 3), and show relatively high $Fe^{3+}\#$ (0.11–0.22) and high Ti (0.15–1.07 wt.% TiO₂, mostly below 0.40 wt.%). These observations suggest that fluid penetration along intragrain fractures caused localized alteration and even homogenization of Cr-spinel. Nevertheless, these compositions remain close to primary and, as discussed more fully in the following text, are markedly different from typical ferritchromite or metamorphosed Cr-spinel.

Clinochlore and magnetite

Well zoned Cr-spinel is commonly partially surrounded by coronae of Mg-rich chlorite, hereafter described as clinochlore (Mg# = 0.94-0.96; Table 2), which contains significant Cr₂O₃ (1.24-3.01 wt.%). The clinochlore is subhedral and bladed in shape, overprints mesh-textured lizardite (Fig. 5*a*), and is commonly intergrown with magnetite near Cr-spinel grains (Fig. 5c,d) suggesting contemporaneous crystallization.

Magnetite commonly occurs at or near rims of all types of Cr-spinel and, like the dusty magnetite grains within the serpentine matrix, has nearendmember compositions ($Fe^{3+}\# = 0.97$; Mg# = 0). The evidence supports its formation during serpentinization, as suggested by Evans and Frost (1975). Such magnetite around zoned Cr-spinel grains forms a distinct phase from the Fe-enriched rims of zoned Cr-spinel (Fig. 5*f*).

Discussion

Origin of zoning: alteration sequence of Crspinel

Textural evidence suggests that the zoning observed in Cr-spinel is not a primary feature; but instead appears to be due to alteration. Indeed, there appears to be a relationship between zoning

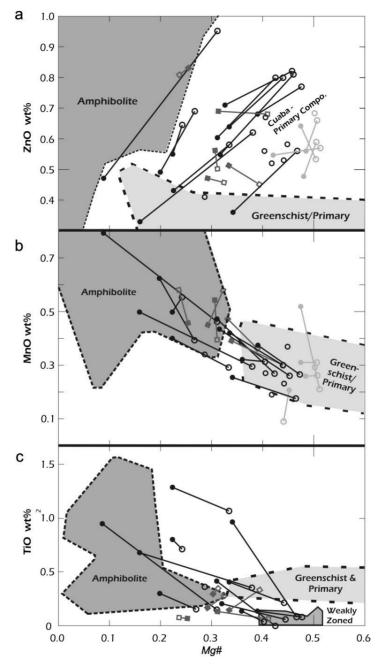


FIG. 8. Minor-element contents vs. Mg# in Cr-spinel. Symbols are as in Fig. 3, with open symbols representing cores (primary compositions), and filled symbols representing rims. (a) ZnO vs. Mg#; (b) MnO vs. Mg#; (c) TiO₂ vs. Mg#. The composition field for Cr-spinel in amphibolite-facies metamorphic rocks is based on data compiled by Barnes (2000) for Cr-spinel in metamorphosed komatiites (see text for discussion). Only those parts of the compositional space relevant to the Cuaba samples are shown. Chromium-spinel which has been metamorphosed under amphibolite-facies conditions commonly contains high Zn (up to ~8.0 wt.% ZnO) and Mn (up to ~2.7 wt.% MnO) which are beyond the values shown in these diagrams.

ZONED CR-SPINEL AND FERRITCHROMITE ALTERATION

	636B-C-7	636B-B-4	636C-1-1	48-tra-19	34A-K-5
SiO ₂	32.48	31.24	32.13	32.42	31.56
$Al_2 \tilde{O}_3$	16.59	17.1	15.98	13.93	17.55
TiO ₂	0.11	0.1	_	_	_
Cr_2O_3	1.24	1.82	2.04	3.01	1.79
FeO	3.06	3.52	3.84	2.79	3.65
MgO	32.6	32.52	33.71	33.59	33.4
CaO	_	0.01	0.01	—	0.01
Cl	0.03	0.02	n.a.	0.04	n.a.
F	0.22	0.17	0.14	0.04	n.a.
Total	86.33	86.5	87.85	85.83	87.97
Formula on the	basis of 28 oxygen a	atoms			
Si	6.18	5.97	6.06	6.24	5.93
Al	1.82	2.03	1.94	1.76	2.07
Sum	8.00	8.00	8.00	8.00	8.00
Al	1.91	1.83	1.62	1.40	1.81
Гі	0.02	0.01	—	—	_
Cr	0.19	0.28	0.31	0.46	0.27
Fe	0.49	0.56	0.61	0.45	0.57
Мg	9.25	9.27	9.48	9.64	9.35
Ca	—	0.00	0.00	—	0.00
Sum	11.85	11.95	12.01	11.95	12.00
CI	0.01	0.01	n.a.	0.02	n.a.
F	0.13	0.10	0.08	0.01	n.a.
Sum	0.14	0.11	0.08	0.04	n.a.
Mg#*	0.95	0.94	0.94	0.96	0.94

TABLE 2. Representative clinochlore (Mg-dominant chlorite) compositions from the Rio San Juan Complex.

* The ratio Mg# = Mg/(Mg + Fe).

The abbreviation n.a. is not analysed; - is below detection limit.

and the degree of hydration and metamorphism of the serpentinite as shown by the following observations: (1) weakly zoned Cr-spinel occurs in sample RD68 which contains primary silicate minerals; (2) apart from sample RD34a, well zoned Cr-spinel occurs in samples which are fully hydrated; (3) grains with abundant fractures occur in samples which show evidence of higher-grade metamorphism, such as the rare cummingtonite, talc and tremolite. Furthermore, the occurrence of clinochlore coronae around Cr-spinel and the texture of clinochlore overprinting lizardite confirm that the zoning in Cr-spinel and the crystallization of clinochlore post-dated serpentinization.

In cumulate rocks, zoning in Cr-spinel might originate as a result of reactions with the intercumulus liquid. However, this mechanism is not applicable to the samples described herein as the serpentinites have a bulk rock geochemistry inconsistent with a cumulate origin, and consistent with an origin as hydrated residual mantle peridotite (Fig. 2; Saumur *et al.*, 2010). There is a possibility that residual mantle peridotites could have reacted with partial melts in the mantle wedge; however, such peridotites would have low Mg contents in the silicate minerals and bulk rocks (e.g. Kelemen *et al.*, 1990; Morgan and Liang, 2003). Therefore, this possibility can be discounted on the basis of the high $Mg^{\#}$ of relict olivine, relict orthopyroxene and bulk rocks. In addition, there is no textural evidence to support this possibility.

We propose the following zoning sequence. The first stage involves the replacement of Mg^{2+}

by Fe^{2+} with constant Fe^{3+} , as observed in 'weakly zoned' Cr-spinel, with possible minor replacement of Al^{3+} by Cr^{3+} . The second stage is illustrated by the formation of well zoned Cr-spinel. The formation, which involves an increase in $Fe^{3+}\#$ at grain rims or along intragrain fractures and the replacement of Al^{3+} by Fe^{3+} . The Al^{3+} lost from the Cr-spinel during the alteration process allows the formation of clinochlore coronae around the Cr-spinel grains. The crystallization of ferritchromite and clinochlore is facilitated by reactions between Crspinel and serpentine or Mg- and Si-rich fluids (e.g. Mellini *et al.*, 2005; Merlini *et al.*, 2009). Iyer *et* al. (2008) suggest the following reactions:

 $Al^{3+} + Serpentine + H_2O \rightarrow Clinochlore$ (2)

By combining these reactions, reaction (3) is obtained, which is consistent with intergrowths of magnetite and clinochlore around Cr-spinel grains, and clinochlore overprinting lizardite:

Cr-spinel + Fe³⁺ + 2H₂O + Serpentine → Ferritchromite + Magnetite + Clinochlore + H₂ (3)

The reactions outlined above do not explain the increases in Cr observed in the rims in both weakly (Fig. 4b-d) and well zoned (e.g. Fig. 6a) Cr-spinel grains. Kimball (1990) showed that during Cr-spinel breakdown, Cr is preferentially incorporated in Cr-spinel and Al in clinochlore, and that this results in higher Cr and lower Al in the remaining Cr-spinel. Although this explanation is consistent with the compositions of the zoned Cr-spinel which have clinochlore coronae, it does not explain the slight increase in Cr# in the rims of weakly zoned Cr-spinel grains which lack marginal clinochlore. The small increase in Cr³⁺ could be partly primary (before serpentinization); for example, it could be related to partial melting which would have preferentially removed Al and consequently increased the Cr content of Cr-spinel in the residue. Alternatively, a reaction involving Cr³⁺ and Al³⁺ exchange between Cr-spinel and lizardite may have increased the Cr content of the Cr-spinel. Lizardite is known to incorporate Al into its structure near to its upper thermal stability limit at ~300°C (Caruso and Chernosky, 1979; Evans, 2004). If Al was sequestered by the surrounding lizardite, this would reduce the Al and increase the Cr content in the outer part of the Cr-spinel.

The alteration sequence in these samples is similar to that reported from the Kalkan Chromitite in the southern Urals by Merlini *et al.* (2009). The serpentine-group mineral in the Kalkan Chromitite is antigorite, which contrasts with the dominance of lizardite in this study. Therefore the polymorphism of the serpentine phase involved in the reaction with Cr-spinel appears to have no bearing on the occurrence of ferritchromite alteration.

Thermal conditions of alteration of Cr-spinel

The formation of ferritchromite in serpentinites requires heating after serpentinization (e.g. Cerny, 1968), and a variety of conditions have been proposed for its onset. Temperatures between 300°C and 400°C during greenschist-facies metamorphism (Standish et al., 2002; Iver et al., 2008; González-Jiminez et al., 2009; Merlini et al., 2009), and as high as 500°C during amphibolite-grade metamorphism (e.g. Bliss and MacLean, 1975; Pinsent and Hirst, 1977; Barnes, 2000; Mellini et al., 2005), have been proposed. Furthermore, different types of reactions have been proposed for the formation of ferritchromite. Divalent cations in Cr-spinel are known to be mobile at low temperatures (Barnes, 2000) and alteration in Cr-spinel similar to that reported here in the weakly zoned samples has been observed in greenschist-facies metamorphic rocks (e.g. Evans and Frost, 1975; Suita and Strieder, 1996; González-Jiménez et al., 2009). The mobility of trivalent cations, notably Fe3+, is favoured by oxidizing conditions, which typically occur in serpentinites during prograde metamorphism after serpentinization (Frost, 1985).

The compositions of the Cr-spinel rims in our study differ from ferritchromite compositions reported from amphibolite grade metamorphic terranes. Indeed, the Fe^{3+} # ratios of 0.03–0.35 at grain rims are low compared to those of typical ferritchromite (usually >0.5, see Fig. 7), and the Zn, Mn and Ti contents are generally lower than ferritchromite in amphibolite-grade rocks (Fig. 8, see discussion below). Such compositions suggest that ferritchromite is not fully developed and is compositionally immature. Furthermore, ferritchromite in amphibolite-grade rocks commonly replaces the entire Cr-spinel grains, whereas in these samples the alteration is, for the most part, limited to the outer rims. Finally, in high metamorphic grade terranes, complete clinochlore coronae are common around Cr-spinel grains, whereas only partial clinochlore coronae have developed in our samples. Thus, compositional and textural evidence suggest that reaction 3 is incomplete, and that the rims of zoned Cr-spinel represent only the initial stages of ferritchromite alteration.

The immature development of ferritchromite is further supported by the rarity of high-temperature minerals in the Cuaba serpentinites. Cummingtonite, tremolite and talc occur only rarely in samples RD34 and RD6-36. Cummingtonite is generally considered to form at high temperatures (e.g. Frost, 1975; Evans, 1977), whereas talc and tremolite, although they form over a wide range of temperatures, generally crystallize above 400°C (Evans and Frost, 1975). Samples containing these minerals also contain Cr-spinel showing extensive microscopic fractures representing the highest degree of Cr-spinel alteration observed in this study. High-temperature phases are notably absent in sample RD68, which shows only subtle zoning in the Cr-spinel. The correlation of the high-temperature minerals and the alteration of Cr-spinel suggests that they are formed by related processes.

Although some high-temperature minerals were identified, the Cuaba serpentinites are dominated by lizardite, which is stable at temperatures of up to ~300°C (Evans, 2004; Schwartz et al., 2013). Lizardite alters to antigorite at higher temperatures, but no antigorite was detected in the samples. Furthermore, no textural evidence for the retrograde replacement of antigorite by lizardite was observed. Lizardite typically exhibits mesh, hourglass and bastitic pseudomorphic textures after primary olivine and orthopyroxene. The occurrence of lizardite with rare hightemperature phases suggests that it was, at some point, metastable. It is probable that the thermal event, with a peak temperature which appears to have been in excess of 300°C, was too short-lived to allow the conversion of lizardite to antigorite.

Therefore, the degree of alteration of Cr-spinel and the overall mineralogy of the Cuaba serpentinites are consistent with heating above the serpentinization temperature, which is consistent with previous work. However, we emphasize that in the case of the Cuaba serpentinites, this thermal event was probably brief, inhibiting the conversion of lizardite to antigorite, resulting in the rarity of high-temperature minerals, and preventing the complete development of ferritchromite and clinochlore coronae.

The major element compositional variations in Cr-spinel documented in this study are similar to

those reported from chromitites in the Dobromirtsi Massif in southeast Bulgaria (González-Jiménez *et al.*, 2009). Those samples show an analogous sequence of alteration, defined by initial increases in Cr^{3+} and Fe^{2+} and associated decreases in Al^{3+} and Mg^{2+} , followed by an increase in Fe^{3+} at grain rims. The first stages of alteration occur at greenschist-facies conditions, and subsequent increases in Fe^{3+} occur at higher grade metamorphic conditions. Although the sequence of alteration is consistent with that described in this study, the compositional changes are greater in extent in Dobromirtsi samples than in Cuaba samples because of sustained high-grade metamorphism under amphibolite facies conditions.

Primary Zn content of Cr-spinel

The Zn contents in Cr-spinel cores from the Cuaba serpentinites range between 0.46 and 0.95 wt.% ZnO (Fig. 8*a*). Previous work has suggested that Zn concentrations in excess of 0.5 wt.% ZnO in Cr-spinel are noteworthy (Groves *et al.*, 1983; Wylie *et al.*, 1987; Johan and Ohnenstetter, 2010, and references therein). They are especially uncommon in forearc mantle peridotites, probably because of the mildly incompatible behaviour of Zn during partial melting of the mantle which is inferred from the slightly higher Zn contents of mid-oceanic ridge basalt and continental crust compared to the primitive mantle value (e.g. McDonough and Sun, 1995).

A high Zn content in Cr-spinel, of 1-8 wt.% ZnO, is common in metamorphosed ultramafic rocks, (e.g. Paktunc and Cabri, 1995; Barnes, 2000; Säntti *et al.*, 2006). However, the general lack of a metamorphic overprint on the Cr-spinel cores in the Cuaba samples suggests that a metamorphic origin for the high Zn contents is unlikely. Interactions with late-stage magmatic fluids, which are known to cause metasomatic Zn-enrichment in Cr-spinel (e.g. Johan and Ohnenstetter, 2010), is also unlikely as the Cuaba serpentinites originated from residual mantle peridotite.

High Zn contents in Cr-spinels in forearc peridotites might be explained by their reequilibration with olivine prior to serpentinization. Zinc has a strong affinity with spinel compared to other silicate minerals. The partition coefficient of Zn between Cr-spinel and olivine is far greater than 10 and strongly dependent on temperature, with higher Zn contents in Cr-spinel expected at lower temperatures (e.g. Ryan *et al.*, 1996; De Hoog *et al.*, 2010). The subduction zone below the Caribbean plate was exceptionally cool: a lawsonite schist was produced in the RSJC and Samaná Peninsula (Zack *et al.*, 2004; Schertl *et al.*, 2012); and the geothermal gradient of the subduction zone is estimated to have been ~8°C km⁻¹ (Zack *et al.*, 2004; Krebs *et al.*, 2008; Schertl *et al.*, 2012). Thus the high Zn contents of the Cr-spinel cores could be attributed to the unusually cool temperatures of the northern Caribbean forearc.

Behaviour of minor elements during Cr-spinel alteration

The compositions of Cr-spinel in the Cuaba samples can be compared to those of spinel in komatiites compiled by Barnes (2000). Using the diagrams presented in that contribution, we have created fields for 'amphibolite' and 'greenschist/ primary' spinel (Fig. 8). The komatiites are variably metamorphosed and the plots illustrate the compositional characteristics of Cr-spinel which has undergone metamorphism. Spinel in amphibolite-facies rocks has consistently high Zn, Mn and Ti contents and lower *Mg#* compared to that in greenschist-facies rocks (Fig. 8).

In the Cuaba samples, the Cr-spinel rims have lower concentrations of Zn (0.32-0.71 wt.% ZnO; Fig. 8a) compared to cores, and their compositions plot in the field for Cr-spinel that has been metamorphosed under greenschist-facies conditions. Compositional variations between cores and rims suggest that Zn behaves like Mg (Fig. 6) and is liberated from Cr-spinel during initial-stage reactions with serpentine or Mg- and Si-rich fluids, as suggested by González-Jiménez et al. (2009). If these samples had undergone amphibolite-facies metamorphism and the duration of this event had been sufficient to alter the compositions of Cr-spinel, an enrichment in Zn would be expected in Cr-spinel, comparable to that observed amphibolite-grade spinels in komatiites (Fig. 8).

The Mn contents of the Cr-spinel cores are low (0.15-0.47 wt.% MnO; Fig. 8b), and the rims show a progressive increase in Mn with increasing total Fe content, with some plotting in, or near, the amphibolite field. The behaviour of Ti is similar to that of Mn (Figs 6b and 8c). Increasing Mn and Ti are most probably the result of alteration of Cr-spinel, consistent with the well documented increases of Mn and Ti in Cr-spinel in amphibolite-grade rocks (Barnes, 2000). The

evidence suggests that Mn and Ti follow the same well documented progressive enrichment during alteration as Cr-spinels reported in metamorphosed komatiites (Fig. 8b,c). Furthermore, our data suggests that they may be enriched in Cr-spinel prior to Zn during prograde metamorphism, although this might depend on the composition of the syn-metamorphic fluid.

Vanadium has a wide range of oxidation states, from +2 to +5. Divalent and trivalent V are compatible with silicate and oxide minerals, but pentavalent V is incompatible (Mallmann and O'Neill, 2009). As mantle wedges are generally oxidized (e.g. Parkinson and Arculus, 1999; Wang et al., 2008), V in forearc and subarc mantle is expected to be incompatible. Consequently, V contents are low in the mantle and in Cr-spinel (Richter et al., 2006). During metamorphism, the V content of Cr-spinel generally increases, as shown by the high $(9 \text{ wt.}\% \text{ as } V_2O_3)$ content in some rocks metamorphosed to amphibolite-facies conditions (Säntti et al., 2006). Although V levels are not as high in the Cuaba samples, the slight increase in V at grain rims (Fig. 6b) is consistent with previous reports (Säntti et al., 2006) of V enrichment during Cr-spinel alteration.

Causes of Cr-spinel zoning in forearc serpentinites

As noted in the introduction, ferritchromite is very rare in forearc mantle peridotites. Our detailed examination of forearc mantle serpentinites in the Marianas and the Himalayas failed to identify ferritchromite (Saumur *et al.*, 2011). Furthermore, ferritchromite is notably absent in forearc serpentinite along the Camú fault which originated from the same forearc mantle as the Cuaba samples, but is closer to the trench than these samples from the Septentrional fault, (Fig. 1; Saumur *et al.*, 2010).

The lack of ferritchromite alteration in typical forearc mantle serpentinites can be explained by thermal histories which differ from that of the Cuaba serpentinites. Forearc mantle serpentinites are usually exhumed from the base of a mantle wedge in subduction channels located along subducting plates (e.g. Gerya *et al.*, 2006), and such exhumation is accompanied by a simple cooling, without heating which would cause these serpentinites to go through amphibolite-facies conditions. The occurrence of ferritchromite alteration in the Cuaba samples suggests that

they have been affected by a late heating event atypical of usual serpentinite exhumation. The reports of amphibolite-facies conditions within Omi forearc serpentinites, which also exhibit ferritchromite in accessory Cr-spinel (Tsujimori *et al.*, 2004) are consistent with this suggestion.

There are two possible explanations for the occurrence of ferritchromite in the Cuaba samples. The first possibility is related to the protrusion of these serpentinites from the base of the mantle wedge towards higher crustal levels along the Septentrional fault zone (Saumur et al., 2010). The interiors of mantle wedges are hotter than their bases near the slab, and therefore Cuaba serpentinites would have been heated during their protrusion through the forearc. This would also explain the lack of ferritchromite in serpentinites along the Camú fault zone: because of their position closer to the palaeo-trench, compared to Cuaba serpentinites, the Camú serpentinites would have protruded through a thinner and cooler mantle wedge than the Cuaba serpentinites, and therefore would not have experienced significant temperature increases during protrusion.

The second possible source of heating is related to the emplacement of the Rio Boba Intrusion, which is located 1-2 km north of Rio Cuevas and Loma Quita Espuela (Fig. 1). Again, this explanation would be consistent with the lack of Cr-spinel zoning in CFZ serpentinites, which are located at a greater distance from the known northern extent of the Rio Boba Intrusion (>15 km), and thus were probably unaffected by its emplacement.

Conclusion

We report the rare occurrence of ferritchromite alteration of Cr-spinel in forearc serpentinites located in the southern part of the RSJC in the northern Dominican Republic. Ferritchromite occurs rarely in hydrated forearc mantle peridotites because they are not commonly heated during exhumation.

Chromium-spinels in this study record the early stages of ferritchromite formation. The first stage of alteration is characterized by the replacement of Mg^{2+} by Fe^{2+} and, to a lesser extent, AI^{3+} by Cr^{3+} . This is followed by the replacement of AI^{3+} by Fe^{3+} at slightly higher temperatures. The Mg and Al (and minor Cr) released from Cr-spinel produce partial coronae of Cr-bearing clinochlore, which is commonly intergrown with magnetite. This alteration occurred at conditions at which

lizardite, the dominant species of serpentinite in these rocks, was metastable, and rare highertemperature minerals such as talc, tremolite and cummingtonite crystallized. Thus, the observed Cr-spinel zoning (i.e. compositionally and texturally immature ferritchromite rims) is interpreted to have formed in response to a short-lived thermal event that did not allow the full development of ferritchromite and clinochlore coronae.

The contents of Ti and Mn in the Cr-spinel rims are consistent with our interpretation that the serpentinites underwent only minor degrees of metamorphism. Relatively high Zn in the cores reflect cool temperatures of the northern Caribbean mantle wedge, whereas lower Zn at rims was produced during the early stages of Crspinel alteration.

This study confirms that the compositions of Cr-spinel may be modified during low-grade metamorphism. Some caution must therefore be exercised in using Cr-spinel compositions for petrogenetic studies, even in slightly metamorphosed serpentinites.

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