

**Abstract, 2019 Joint Annual Meeting of Geological Association of Canada and
Mineralogical Association of Canada in Quebec City.**

**The occurrence and chemistry of Ti-oxide phases and their timing of
crystallization along the P2 fault, Athabasca Basin, Saskatchewan, Canada**

Abstract No.: SS-RE13-O12

*Session/Symposia: Uranium ore systems research: Fluid pathways, metal traps and expressions
of the systems*

Authors:

*Erin **Adlakha**, Saint Mary's University*

*Keiko **Hattori**, University of Ottawa*

*Mitchell **Kerr**, Saint Mary's University*

*Brandon **Boucher**, University of New Brunswick*

Titanium oxide minerals are examined in hydrothermally altered rocks along the P2 fault, the main structure hosting the McArthur River deposit. Two TiO₂ polymorphs occur with contrasting mineral assemblages: i) rutile with oxy-dravite, graphite, pyrite, zircon and quartz in basement metapelite, and ii) anatase with hematite ($\pm$ dolomite, kaolin, illite and aluminum phosphate sulfate minerals) in basement rocks and overlying sandstone. Rutile contains variably high Nb₂O₅ (up to 3.86 wt.%), Ta₂O₅ (up to 0.42 wt.%), Cr₂O₃ (up to 0.48 wt.%), Fe₂O₃ (up to 1.48 wt.%), ZrO₂ (up to 0.49 wt.%), WO₃ (up to 2.54 wt.%), and V₂O₃ (up to 2.12 wt.%), suggesting low water/rock ratios during crystallization. The coupled substitution mechanisms $M^{3+} + M^{5+} \rightarrow 2Ti^{4+}$ and $2M^{3+} + M^{6+} \rightarrow 3Ti^{4+}$ partially explain the incorporation of non-tetravalent cations in rutile. However, high abundances of trivalent cations relative to pentavalent and hexavalent cations, coupled with weak Raman shifts near 3100 cm⁻¹, indicate protonation of oxygen in rutile crystal structure through the exchange $M^{3+} + OH^- \rightarrow Ti^{4+} + O^{2-}$. The Zr-in-rutile geothermometer yields temperatures between 740 and 890 °C, reflecting granulite facies regional metamorphic conditions of the 1.8 Ga Trans Hudson Orogeny. However, the U-Pb ages of rutile are young, ranging from 1726 to 1771 Ma. The rutile ages represent either i) slow cooling to 400-500 °C, or ii) a thermal event, possibly related to the ~1.75 Ga anorogenic Neultin Suite. With the exception of variably high Fe₂O₃ (up to 1.96 wt.%), anatase contains low elemental impurities reflecting crystallization at low temperature. High Fe³⁺ is likely accommodated by substitution with OH⁻. Anatase gives a relatively young age of 1569 $\pm$ 31 Ma in the basement far below the unconformity, indicating the onset of oxidizing basinal hydrothermal activity in the basement rocks along the P2 fault.